Cramer's Theorem in Probability and Statistical Mechanics An Exercise in Sign Shenanigans

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1 Introduction

The purpose of this document is to figure out exactly what is going on with Cramer's theorem, and the difference in sign conventions of the partition function in probability and in statistical mechanics.

2 Cramer's Theorem

From [2] we have the following statement of Cramer's theorem

THEOREM 1. (CRAMER'S THEOREM) Let $X_1, X_2, ...$ be i.i.d. random variables with finite logarithmic moment generating function

$$\Lambda(t) \coloneqq \log \mathbb{E}[e^{tX_1}]$$

Let

$$\Lambda^*(x) \coloneqq \sup_{t \in \mathbb{R}} (t x - \Lambda(t))$$

be the Legendre transform of Λ . Then for every $x \ge \mathbb{E}[X_1]$,

$$\lim_{n\to\infty}\frac{1}{n}\log\mathbb{P}\left[\sum_{j=1}^nX_j\geq x\,n\right]=-\Lambda^*(x)$$

In statistical mechanics, we use a different convention for our "partition function"; we define

$$\Lambda(\beta) := \log \mathbb{E}[e^{-\beta H}]$$

where H is a random variable giving the energy of the system. We will show how to adopt Cramer's theorem to this situation.

Let $Y_1, Y_2,...$ be i.i.d. random variables distributed as H, and let $X_i = -Y_i$. Then we can apply Cramer's theorem to the X_i to get that for every $x \ge \mathbb{E}[X_1]$,

$$\lim_{n\to\infty}\frac{1}{n}\log\mathbb{P}\left[\sum_{j=1}^nX_j\geq x\,n\right]=-\Lambda^*(x)$$

Now, suppose that $x \le \mathbb{E}[Y_1]$, so that $-x \ge \mathbb{E}[X_1]$. Then

$$\mathbb{P}\left[\sum_{j=1}^{n} Y_{j} \leq x \, n\right] = \mathbb{P}\left[\sum_{j=1}^{n} X_{j} \geq (-x) \, n\right]$$

Thus,

$$\lim_{n\to\infty}\frac{1}{n}\log\mathbb{P}\left[\sum_{j=1}^nY_j\leq x\,n\right]=-\Lambda^*(-x)$$

If we expand out the definition of $-\Lambda^*(-x)$, we get

$$-\Lambda^*(-x) = -\sup_{t \in \mathbb{R}} (t(-x) - \Lambda(t))$$
$$= -\sup_{t \in \mathbb{R}} (-tx - \Lambda(t))$$
$$= \inf_{t \in \mathbb{R}} (tx + \Lambda(t))$$

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as when we move the negative sign across the supremum it becomes an infimum. Thus, we can state an alternative, statistical mechanical version of Cramer's thereom, adding in a purely cosmetic replacement of t with β and x with u. This is the version that [1] uses.

THEOREM 2. (CRAMER'S THEOREM, STATISTICAL MECHANICAL VERSION) Suppose that $Y_1, Y_2,...$ are i.i.d. random variables, all distributed as H. Let $\Lambda(\beta)$ be the logarithmic partition function, defined by

$$\Lambda(\beta) = \log \mathbb{E}[e^{-\beta H}]$$

Define the (not quite Legendre transform of Λ) by

$$\Lambda^*(u) = \inf_{\beta \in \mathbb{R}} (\beta u + \Lambda(\beta))$$

Then, for every $u < \mathbb{E}[H]$,

$$\lim_{n\to\infty}\frac{1}{n}\log\mathbb{P}\left[\sum_{j=1}^nY_j\leq u\,n\right]=\Lambda^*(u)$$

3 Relation to Classical Thermodynamics

We now try and investigate exactly what Λ and Λ^* mean in the context of classical thermodynamics. Let \mathfrak{X} be a countable set, and let $H:\mathfrak{X}\to\mathbb{R}$ be any function. Then let X be a variable taking values in \mathfrak{X} , with distribution

$$\mathbb{P}(X=i) = \frac{e^{-\beta H(i)}}{Z(\beta)}$$

where

$$Z(\beta) = \sum_{i \in \mathfrak{X}} e^{-\beta H(i)}$$

is called the partition function. Surprisingly, the log of partition function,

$$\Lambda(\beta) = \log Z(\beta)$$

can be used to calculate quantities related to the system fairly easily. For instance,

$$\Lambda'(\beta) = \frac{\partial}{\partial \beta} \log Z(\beta)$$

$$= \frac{\sum_{i \in \mathfrak{X}} \frac{\partial}{\partial \beta} e^{-\beta H(i)}}{Z(\beta)}$$

$$= \sum_{i \in \mathfrak{X}} -H(i) \frac{e^{-\beta H(i)}}{Z(\beta)}$$

$$= -\sum_{i \in \mathfrak{X}} H(i) \mathbb{P}(X = i)$$

$$= -\mathbb{E}[H(X)]$$

In physics, we often write this last quantity as $-\langle H \rangle$.

We can also calculate the entropy using the partition function

$$S = -\sum_{i \in \mathfrak{X}} P_i \log P_i$$

$$= -\sum_{i \in \mathfrak{X}} \frac{e^{-\beta H(i)}}{Z(\beta)} \log \left(\frac{e^{-\beta H(i)}}{Z(\beta)} \right)$$

$$= -\sum_{i \in \mathfrak{X}} -\beta H(i) P_i + \sum_{i \in \mathfrak{X}} P_i \log(Z(\beta))$$

$$= \beta \langle H \rangle + \log(Z(\beta))$$

We can now see if this matches up with the classical definition of entropy for the Gibbs ensemble. Classically, entropy is given by the formula

$$F = U - TS$$

where F is the "free energy". We want to identify F with $-\frac{\log(Z(\beta))}{\beta} = -T\log(Z(\beta))$, let's see if that ends up agreeing with our calculation for entropy above.

$$S = \beta \langle H \rangle + \log(Z(\beta))$$

$$TS = \langle H \rangle + T \log(Z(\beta))$$

$$TS = \langle H \rangle - F$$

$$F = \langle H \rangle - TS$$

It does!

BIBLIOGRAPHY

- [1] Jeffrey Commons, Ying-Jen Yang, and Hong Qian. Duality symmetry, two entropy functions, and an eigenvalue problem in gibbs' theory. 2021.
- [2] Achim Klenke. Probability Theory: A Comprehensive Course. Springer-Verlag, London, 2014.